Roll No.

## **PHY-502**

## **Statistical Mechanics and Quantum Mechanics**

M. Sc. PHYSICS (MSCPHY–12/13/16)

First Year, Examination, 2017

Time : 3 Hours

### Max. Marks: 70

Note: This paper is of seventy (70) marks containing three (03) sections A, B and C. Attempt the questions contained in these sections according to the detailed instructions given therein.

#### Section-A

### (Long Answer Type Questions)

- **Note :** Section 'A' contains four (04) long answer type questions of fifteen (15) marks each. Learners are required to answer *two* (02) questions only.
- 1. Explain postulates of statistical mechanics. Determine the number of accessible states for a macro system. 15
- Write the basic assumptions of Maxwell-Boltzmann statistics and establish the necessary formula for it. Explain its significance.
- 3. (a) What are eigen values and eigen functions ? 5
  - (b) Calculate the expectation value of p for the normalized wave function  $\psi x = \sqrt{\frac{2}{L}} \sin \frac{\pi x}{L}$  in

the region 0 < x < L and  $\psi x = 0$  outside this region. 5

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(c) The wave function  $\psi$  of a particle is given by

$$\psi = N \exp\left(\frac{-x^2}{2\beta}\right)$$
 for  $-\infty < x < \infty$ . Find the value of N. 5

value of N.

Explain Klein-Gordon relativistic equation. Give its 4. covariant form. Discuss some applications of Klein-Gordon equation. 15

#### Section-B

#### (Short Answer Type Questions)

- **Note :** Section 'B' contains eight (08) short answer type questions of five (5) marks each. Learners are required to answer six (06) questions only.
- Explain the difference between Schrödinger picture 1. and Heisenberg picture.
- Discuss variational method of finding approximate 2. ground energies and wave functions.
- How was quantum mechanics evolved ? Explain. 3.
- Describe the basic postulates of quantum mechanics. 4.
- Obtain Schrödinger wave function for a particle in a 5. square well potential.
- Explain spin angular momentum. Give its significance. 6.
- Explain the position function for a canonical ensemble. 7.

8. Show that the operator 
$$\left(\stackrel{\rightarrow}{\sigma_1}, \stackrel{\rightarrow}{\sigma_2}\right)^n$$
 for two particles,

each of spin  $-\frac{1}{2}$ , can be linearly expressed in terms of  $\sigma_1, \sigma_2$ .

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### Section-C

#### (Objective Type Questions)

- **Note :** Section 'C' contains ten (10) objective type questions of one (01) mark each. All the questions of this section are compulsory.
- 1. The total number of microstates for the distribution of 5 particles in two similar boxes is :
  - (a) 5<sup>2</sup>
  - (b) 2<sup>5</sup>
  - (c)  $2^{-5}$
  - (d) 5
- 2. In a canonical ensemble the same quantities are :
  - (a) temperatures, volume and number of particles
  - (b) energy of particles
  - (c) energy, volume and number of particles
  - (d) volume and number of particles
- 3. The entropy S and partition function of an ideal gas are related as :
  - (a)  $S = Nk \log_e z$
  - (b)  $S = Nk \log_e \left\{ z + \frac{3}{2} \right\}$

(c) 
$$S = Nk \log_e z + \frac{3}{2} Nk$$

(d) 
$$S = Nk \log_e z + \frac{3}{2}$$

- 4. In the quantum statistics the smallest particle of the system is :
  - (a) molecule
  - (b) atom
  - (c) electron
  - (d) None of these
- 5. Operator form of time dependent Schrödinger equation is :
  - (a)  $H\psi = 1$
  - (b)  $H\psi = A$
  - (c) HA = AH
  - (d)  $H\psi = E\psi$
- 6. The eigen values of Hermitian operator are always :
  - (a) real
  - (b) imaginary
  - (c) complex
  - (d) orthogonal
- 7. The potential function of a harmonic oscillator is :
  - (a) linear
  - (b) elliptical
  - (c) hyperbolic
  - (d) parabolic
- 8. The potential energy of a particle in one-dimensional box is :
  - (a) zero
  - (b) ∞
  - (c) 1
  - (d) None of these

- 9. The correct relation between shift operators is :
  - (a)  $\sigma_+ = \sigma_x + i \sigma_y$
  - (b)  $\sigma_+ = \sigma_x i\sigma_y$
  - (c)  $\sigma_{-} = \sigma_{y} i\sigma_{x}$
  - (d)  $\sigma_{-} = \sigma_{x} + i\sigma_{y}$
- 10. In the time independent perturbation approach, the Hamiltonian operator H of the system is written as :
  - (a)  $H = H^0 + H^1$
  - (b)  $H = H^0 H^1$
  - (c)  $H = H^0 + 3 H^1$
  - (d) None of these

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