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Roll No. -----

MCH-503

Physical Chemistry-1

M.Sc. Chemistry (MSCCH-20)

1st Semester Examination June 2022

Time: 2 Hours

Max. Marks: 40

Note : This paper is of Forty (40) marks divided into two (02) Sections A and B. Attempt the questions contained in these sections according to the detailed instructions given therein.

Section – A

(Long Answer – type questions)

Note: Section 'A' contains Five (05) long-answer-type questions of Ten (10) marks each. Learners are required to answer any two (02) questions only.

[2 x 10 = 20]

- Q.1. (a) What is Hamiltonian operator?
(b) Discuss the important postulates of Quantum mechanism.
- Q.2. Derive the Hamiltonian operator for a one dimensional Harmonic Oscillator.

P.T.O.

- Q.3. (a) Discuss the shape of various orbitals.
(b) What are Quantum number? Give its types.
- Q.4. Describe the first order perturbation theory.
- Q.5. What is Born-oppenheimer approximation?

Section – B

(Short-answer-type questions)

Note: Section 'B' contains Eight (08) short-answer-type questions of Five (05) marks each. Learners are required to answer any Four (04) questions only.

[4 x 5 = 20]

- Q.1. States the second law of thermodynamics in its various forms.
- Q.2. What do you understand the term "Fugacity"?
- Q.3. What are ideal and non-ideal solutions? Give their examples.
- Q.4. Write a short note on
(a) Nernst Heat Theorem
(b) Third law of Thermodynamics

P.T.O.

- Q.5. Give the derivation of Schrodinger Wave equation.
- Q.6. Give the solution of wave equation for a particle in one dimensional box.
- Q.7. Give the solution of wave equation for Hydrogen like atom.
- Q.8. Give the MO theory of H_2^+ ion.
