S-470

Total Pages: 4 Roll No.

MSCCH-604

Photo Chemistry and Allied Chemistry

M.Sc. Chemistry (MSCCH)

3rd Semester Examination, 2022 (Dec.)

Time: 2 Hours] Max. Marks: 70

Note: This paper is of Seventy (70) marks divided into two (02) Sections A and B. Attempt the questions contained in these sections according to the detailed instructions given therein.

SECTION-A (Long Answer Type Questions)

Note: Section 'A' contains Five (05) long answer type questions of Nineteen (19) marks each. Learners are required to answer any Two (02) questions only.

 $(2 \times 19 = 38)$

- **1.** Write notes on the following :
 - (a) Di- π -methane rearrangement.
 - (b) Photosensitization.

- **2.** Discuss photoisomerisation reactions involving formation of Prismane and Dewar benzene as reaction intermediates.
- **3.** Write the major product of the following reactions :

(d)
$$\downarrow$$
 COOCH₃ + \uparrow \uparrow \uparrow \uparrow \uparrow \uparrow \uparrow \uparrow

- **4.** Write notes on the following:
 - (a) Chemiluminiscence.
 - (b) Concerted [2+2] cycloaddition.
- **5.** Suggest mechanism for the given reactions :

(a)
$$\frac{hv}{CH_3OH}$$
 COOCH₃

SECTION-B

(Short Answer Type Questions)

Note: Section 'B' contains Eight (08) short answer type questions of Eight (08) marks each. Learners are required to answer any Four (04) questions only. (4×8=32)

- **1.** Explain the phenomenon of fluorescence and phosphorescence.
- **2.** Discuss the photochemistry of cis-trans isomerization of alkenes.
- **3.** Write notes in the following :
 - (a) Primary and secondary process.
 - (b) Frank-Condon Principle.
- **4.** Explain Norrish type I and type II reactions with suitable examples.
- **5.** Discuss the mechanism of Hoffmann-Loffler-Freytag reaction.

- **6.** What are the basic principles of Green Chemistry?
- **7.** Define quantum efficiency of a photochemical reaction and explain the reason for low and high quantum yield.
- **8.** Write the mechanism of the following transformation :

