Total Pages : 3

Roll No.

MCH-608

Heterocyclic Compounds and Spectroscopy-III

M.Sc. Chemistry (MSCCH)

4th Semester Examination, 2022 (Dec.)

Time : 2 Hours]

[Max. Marks : 35

Note : This paper is of Thirty Five (35) marks divided into two (02) Sections A and B. Attempt the questions contained in these sections according to the detailed instructions given therein.

SECTION-A (Long Answer Type Questions)

- Note : Section 'A' contains Five (05) long answer type questions of Nine and Half (9¹/₂) marks each. Learners are required to answer any Two (02) questions only. (2×9¹/₂=19)
- 1. What are heterocyclic compounds? Write a chemical reaction for the preparation of any two heterocyclic compounds containing two heteroatoms.

S-457 / MCH-608

- **2.** Write a chemical reaction for the preparation of any two heterocyclic compounds containing four heteroatoms.
- 3. Discuss the structure and aromaticity of Furan.
- 4. Write brief notes on the following :
 - (a) Spin-spin coupling.
 - (b) Coupling constant.
 - (c) Chemical shift.
- **5.** Write any four chemical reactions for the synthesis of oxetanes.

SECTION-B

(Short Answer Type Questions)

- **Note :** Section 'B' contains Eight (08) short answer type questions of Four (04) marks each. Learners are required to answer any Four (04) questions only. (4×4=16)
- **1.** What do you know about magnetically non-equivalent protons?
- 2. Write a short note on Optical Rotatory Dispersion (ORD).

S-457/MCH-608 [2]

- **3.** An aromatic compound (Molecular mass = 135) gives the following signals in its PMR spectrum :
 - (a) Singlet (2.09 δ), 3H
 - (b) A distorted singlet (3.09δ) , 1H
 - (c) A multiplet (7.27 δ), 3H
 - (d) A multiplet (7.75 δ), 2H

Predict the structure of the compound.

- 4. Explain shielding and the deshielding effects with examples.
- 5. Write a short note on 19 F-NMR and 13 C-NMR spectroscopy.
- **6.** Why ¹³C-NMR have greater chemical shift value (in ppm) in compare to ¹H–NMR, explain.
- 7. What are the positive and negative Cotton effects?
- 8. Explain the chemical properties of pyrimidine and xanthine.