Total Pages : 3

Roll No.

MCH-606

Photochemistry & Stereochemistry

M.Sc. Chemistry (MSCCH) 4th Semester Examination, 2022 (Dec.)

Time : 2 Hours]

[Max. Marks : 35

Note : This paper is of Thirty Five (35) marks divided into two (02) Sections A and B. Attempt the questions contained in these sections according to the detailed instructions given therein.

SECTION-A (Long Answer Type Questions)

- Note : Section 'A' contains Five (05) long answer type questions of Nine and Half (9½) marks each. Learners are required to answer any Two (02) questions only. (2×9½=19)
- 1. Give mechanism of norish type 1 process. How many types of carbonyl compounds give this reaction give one example for each.

S-455 / MCH-606

[P.T.O.

- **2.** Discuss the mechanism of the photo reduction of benzophenone leading the formation of Benzpinacol.
- 3. Explain photochemistry of alkenes and dienes in detail.
- **4.** (a) Discuss stereochemistry of E1 reaction.
 - (b) Write a note on the conformation of mono substituted cyclohexane.
- 5. Alkenes give two types of photo dimerization :
 - (a) Concerted [2 + 2] cyclo addition.
 - (b) Nonconcerted cyclo addition.

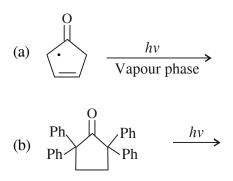
Give mechanism of both these reactions.

SECTION-B

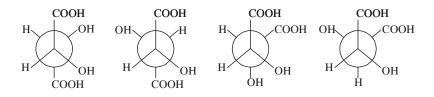
(Short Answer Type Questions)

- **Note :** Section 'B' contains Eight (08) short answer type questions of Four (04) marks each. Learners are required to answer any Four (04) questions only. (4×4=16)
- **1.** Explain fluorescence and phosphorescence with suitable example.
- **2.** Discuss photochemistry of 1,3 -butadiene.
- **3.** Draw the most stable conformation of cis -1-tert-butyl -4 methyl cyclohexane.
- S-455/MCH-606 [2]

- **4.** Explain the effect of conformation on the reactivity of organic compound.
- 5. Complete the following reactions :



6. Comment on the stereochemical and identity aspect of the following Newman projection of tartaric acid



- 7. With suitable example discuss the effect of angle strain and torsional strain on the stability of conformation.
- **8.** Give mechanism of intramolecular hydrogen abstruction reaction given by carbonyl compounds.

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