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## **CHE-551**

# Reaction Mechanisms, Pericyclic Reactions, Photochemistry and Stereochemistry

M. Sc. CHEMISTRY (MSCCH-12/13/16)

Second Year, Examination, 2017

Time: 3 Hours Max. Marks: 80

Note: This paper is of eighty (80) marks containing three (03) Sections A, B and C. Attempt the questions contained in these Sections according to the detailed instructions given therein.

#### Section-A

## (Long Answer Type Questions)

**Note:** Section 'A' contains four (04) long answer type questions of nineteen (19) marks each. Learners are required to answer *two* (02) questions only.

- 1. (a) Describe hemolytic and heterolytic fission of covalent bonds. How do these lead to the formation of carbonium ion and carbon ions? 6
  - (b) Diel's-Alder reaction is a thermally allowed process. Explain with the help of correlation diagram and FMO method.
  - (c) Why is tertiary carbonium ion more stable than primary carbonium ion?

B-22 **P. T. O.** 

[2] CHE-551

- 2. (a) Explain the mechanism of Norrish type-I and Norrish type-II with suitable example. 7
  - (b) Explain electrocyclic reactions with suitable examples. 6
  - (c) Explain the different conformation of cycloherane. 6
- 3. Explain molecular rearrangement reaction. Give their classification and explain the mechanism of pinacolpinacolone and benzilic acid rearrangements.
- 4. Explain the following:
  - (a) Why staggered conformation of ethane is more stable than eclipsed conformation?
  - (b) Mechanism of  $E_2$  and  $E_1$  Cb reaction in detail. 7
  - (c) Jablonski diagram.

## 7

#### Section-B

### (Short Answer Type Questions)

**Note:** Section 'B' contains eight (08) short answer type questions of eight (08) marks each. Learners are required to answer *four* (04) questions only.

- 1. Give *two* methods each for the generation of the following:
  - (a) Carbocations
  - (b) Carbanions
  - (c) Free radical
  - (d) Carbenes
- 2. Describe the conformation of ethane in detail.
- 3. What is cis-trans photoisomerisation?

- 4. Explain symmetry in 2, 4-pentadienyl systems.
- 5. Explain the photochemistry of peroxides and diazocompounds.
- 6. Explain the following:
  - (a) Backmann rearrangement
  - (b) Axial and equatorial bonds in cyclohexane
- 7. Explain the conformation of monosubstituted cyclohexane and disubstituted cyclohaxane.
- 8. With the help of correlation diagram analyse the cis 3, 4-dimethyl cyclobutane  $\rightleftharpoons$  2, 4-hexadiene system, giving stereochemistry under photochemical conditions.

#### Section-C

## (Objective Type Questions)

**Note:** Section 'C' contains ten (10) objective type questions of one (01) mark each. All the questions of this Section are compulsory.

- 1. Reaction intermediate of E<sub>1</sub>Cb reaction is:
  - (a) Carbocation
  - (b) Free radical
  - (c) Carbanion
  - (d) Benzyne
- 2. Geometry of methyl free radical is:
  - (a) Planar
  - (b) Pyramidal
  - (c) Tetrahydral
  - (d) Linear

B-22 **P. T. O.** 

- 3. Hybridization of Alkyl carbanion is:
  - (a) *sp*
  - (b)  $sp^2$
  - (c)  $sp^3$
  - (d)  $sp^2d$
- 4. In which of the following reaction intermediate in carbon?
  - (a) Pinacole-Pinacolone
  - (b) Stevens
  - (c) Claisen
  - (d) Wolf
- 5. Staggered conformation of ethane is more stable than eclipsed conformation. (True/False)
- 6. Tertiary carbonium ion is more stable than primary carbonium ion. (True/False)
- 7. The product obtained in Claisen rearrangement is ........



- 9. Singlet methylene is more stable them triplet methylene reaction. (True/False)

(Fill in the blanks)

#### CHE-551