

To study photoelectric effect

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# Aim, Objectives

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understand the phenomenon Photoelectric effect as a whole.

draw a graph between photocurrent and applied potential for various intensities at constant frequency.

draw a graph between photocurrent and applied potential for various frequencies at constant intensity.

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## Apparatus used, virtual experiment link

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### Physical experiment:

metal plate, Ammeter, Voltmeter, variable power supply, light source s of different powers (Watts) and frequency (Colour and UV light) ,

### Virtual experiment:

<http://vlab.amrita.edu/index.php?sub=1&brch=195&sim=840&cnt=>

needs free sign up and flash player to access the simulation on photoelectric effect virtual experiment

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# Theory: Photoelectric effect

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## Development

Newton corpuscular theory

Huygens's wave theory

-supported by Interference, Diffraction, Polarization etc phenomena.

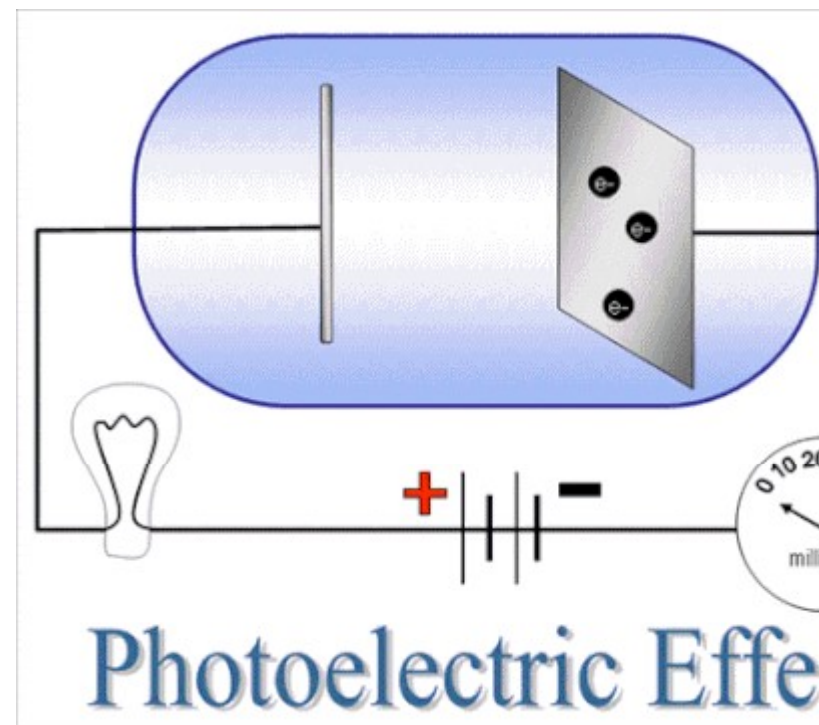
Maxwell's electromagnetic theory

H. Hertz experiment (1886)- photoelectric effect

# Theory: Photoelectric effect

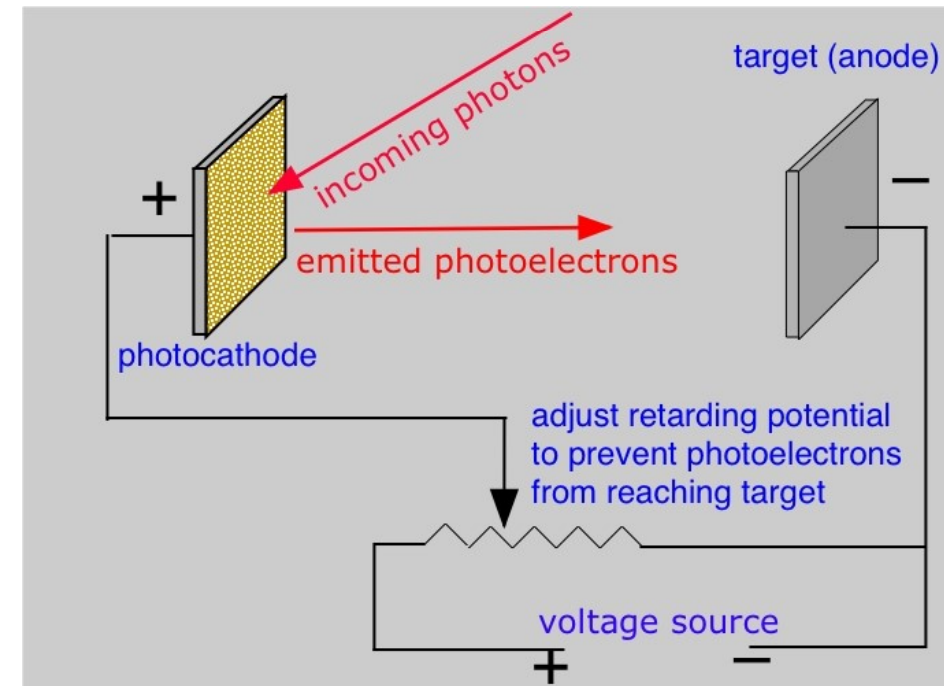
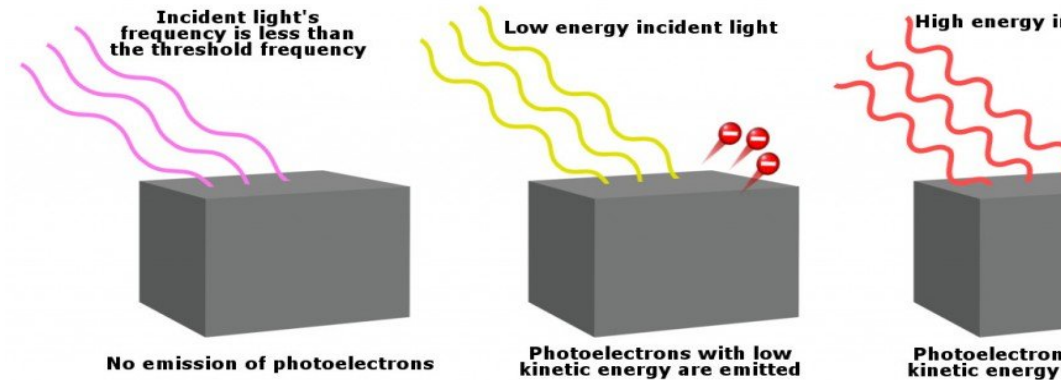
## Development

- When ultraviolet light is allowed to fall on metal plate, electrons are ejected.
- These electrons are known as photoelectrons and current is known as photocurrent.
- This phenomenon is known as photoelectric effect

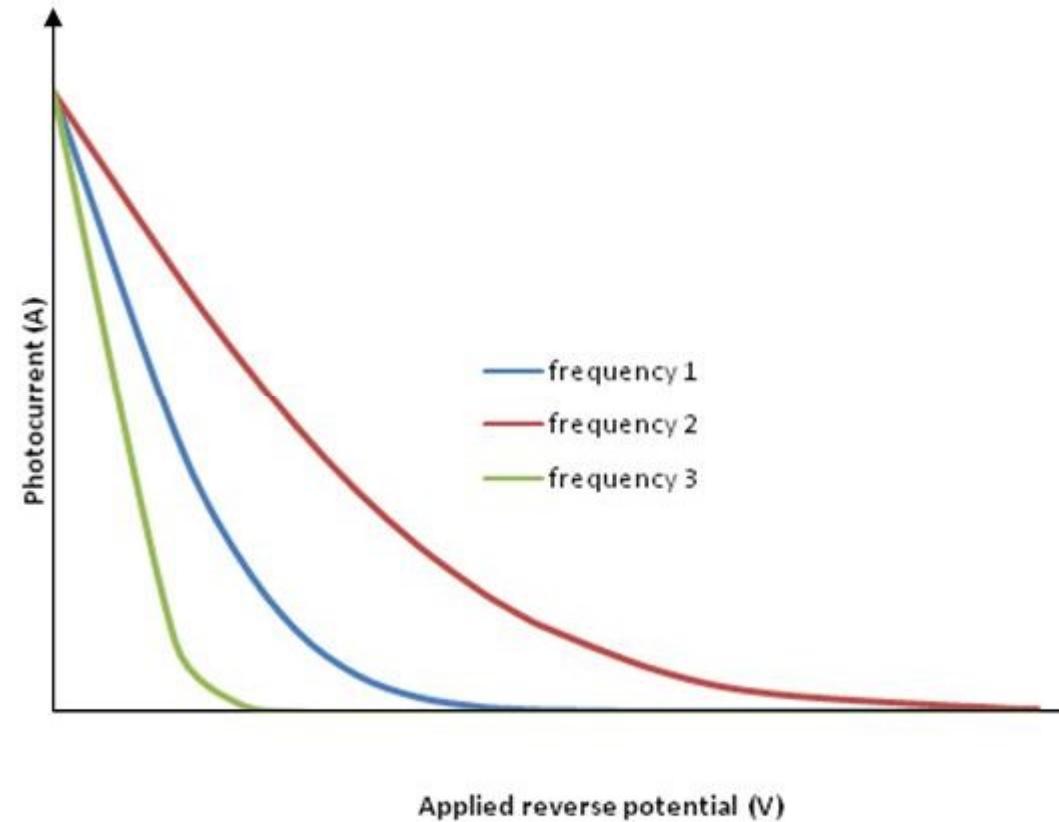
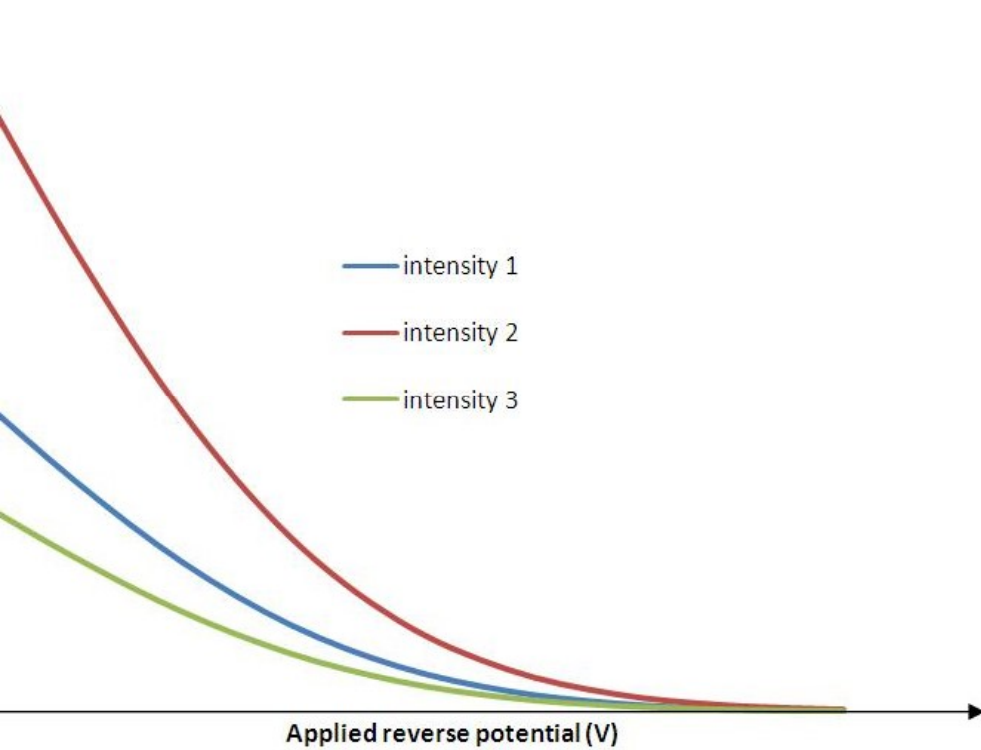


# Theory: Photoelectric effect- experimental observations

- 1] No time lag.
- 2] On increasing intensity at constant frequency, photocurrent increases.
- 3] On frequency variation at constant intensity, stopping potential increases (more negative potential requires to stop photocurrent)



# Theory: Photoelectric effect- experimental observations



# Theory: Photoelectric effect- wave theory fails to explain

- 1] No time lag. –Not explained
- 2] On increasing intensity at constant frequency, photocurrent increases. –Not explained as for wave intensity is related to energy =  $\frac{1}{2} eE^2$  ;  $E = E_0 \sin \omega t$
- 3] On frequency variation at constant intensity, stopping potential increases (more negative potential requires to stop photocurrent) Not explained as for wave frequency is related to no of photoelectrons



# Theory: Photoelectric effect- Einstein particle approach (1905)

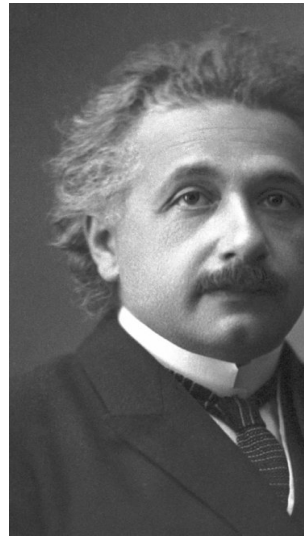
- Light consist of bundle of energy known as photon or quanta ( $E=hf$ ).
- The energy required to emit electron from metal surface- Work function ( $\varphi =hf_o$ ).
- Remaining part will be the KE of the emitted electron.

$$KE= hf - \varphi$$
$$KE= hf - hf_o$$

- When the negative potential on photoelectron receiving plate equal to the KE of emitted electron. No photocurrent will flow in the circuit.

$$KE= eV_o$$

**Nobel prize (1921)**



# Theory: Photoelectric effect- photon theory explains

- 1] No time lag. – explained (particle-particle interaction)
- 2] On increasing intensity at constant frequency, photocurrent increases. – explained as intensity is related to number of photons per unit area.
- 3] On frequency variation at constant intensity, stopping potential increases (more negative potential requires to stop photocurrent) explained as frequency is related to KE of emitted electrons.

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## Procedure: virtual lab

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Go to the given link and login-

<http://vlab.amrita.edu/index.php?sub=1&brch=195&sim=840&cnt=1>

1. Select the material for studying photoelectric effect.

2. Select area of the material, wave-length, intensity of incident light.

3. Switch on the light source.

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## Procedure: virtual lab

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1. Measure the reverse current for various reverse voltages.
2. Plot the current-voltage graph and determine the threshold voltage.
3. Repeat the experiment by varying the intensity for a particular wavelength of incident light.
4. Repeat the experiment by varying the wavelength for a particular intensity of the incident light.

# Observations

Table for intensity variation at constant frequency

Material-

Plate area-

Wavelength=

S.No.	Voltage (V)	Current ( $\mu\text{A}$ )		
		$I=5 \text{ w/m}^2$	$I=10 \text{ w/m}^2$	$I=15 \text{ w/m}^2$
1				
2				
3				
4				
5				
6				
7				
8				
9				
10				
11				
12				

# Observations

Table for intensity variation at constant frequency

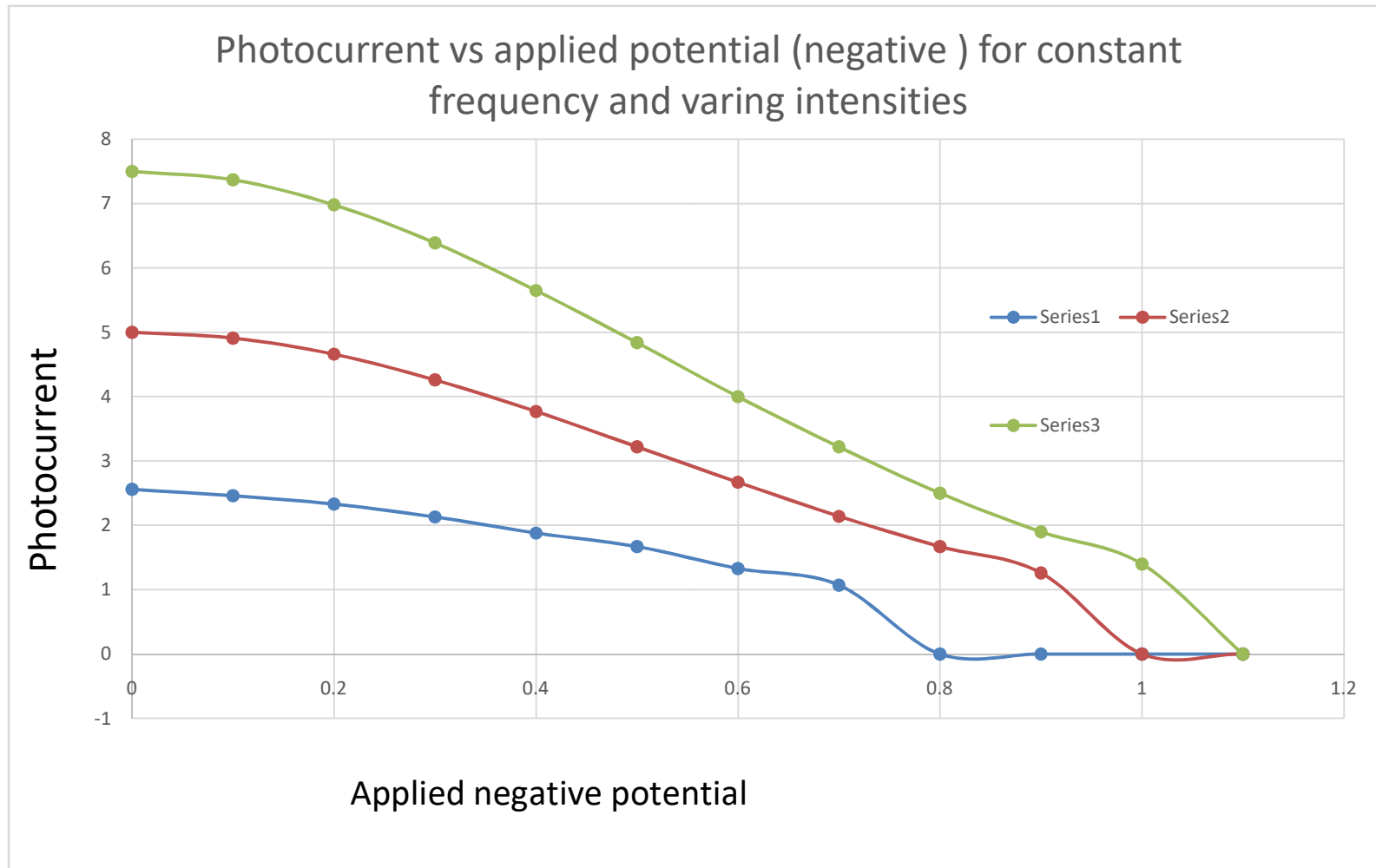
Material- Sodium

Plate area-0.1 cm<sup>2</sup>

Wavelength= 350 nm

S.No.	Voltage (V)	Current ( $\mu$ A)		
		I=5 w/m <sup>2</sup>	I=10w/m <sup>2</sup>	I=15 w/m <sup>2</sup>
1	0	2.56	5	7.5
2	0.1	2.46	4.91	7.37
3	0.2	2.33	4.66	6.98
4	0.3	2.13	4.26	6.39
5	0.4	1.88	3.77	5.65
6	0.5	1.67	3.22	4.84
7	0.6	1.33	2.67	4
8	0.7	1.07	2.14	3.22
9	0.8	0	1.67	2.5
10	0.9	0	1.26	1.9
11	1	0	0	1.4
12	1.1	0	0	0

# Observations- Graph



# Observations

Table for frequency variation at constant intensity

Material-

Plate area-

Intensity=

S.No.	Voltage (V)	Current ( $\mu\text{A}$ )		
		$f=6.64 \times 10^{14} \text{ Hz}$	$f=7.5 \times 10^{14} \text{ Hz}$	$f=8.57 \times 10^{14} \text{ Hz}$
1	0	5	5	5
2	0.1	4.38	4.79	4.91
3	0.2	2.98	4.23	4.66
4	0.3	1.6	3.44	4.26
5	0.4	0	2.59	3.77
6	0.5	0	1.81	3.22
7	0.6	0	1.18	2.67
8	0.7	0	0	2.14
9	0.8	0	0	1.67
10	0.9	0	0	1.26
11	1	0	0	0
12	1.1	0	0	0
13	1.2	0	0	0



# Observations

## Table for frequency variation at constant intensity

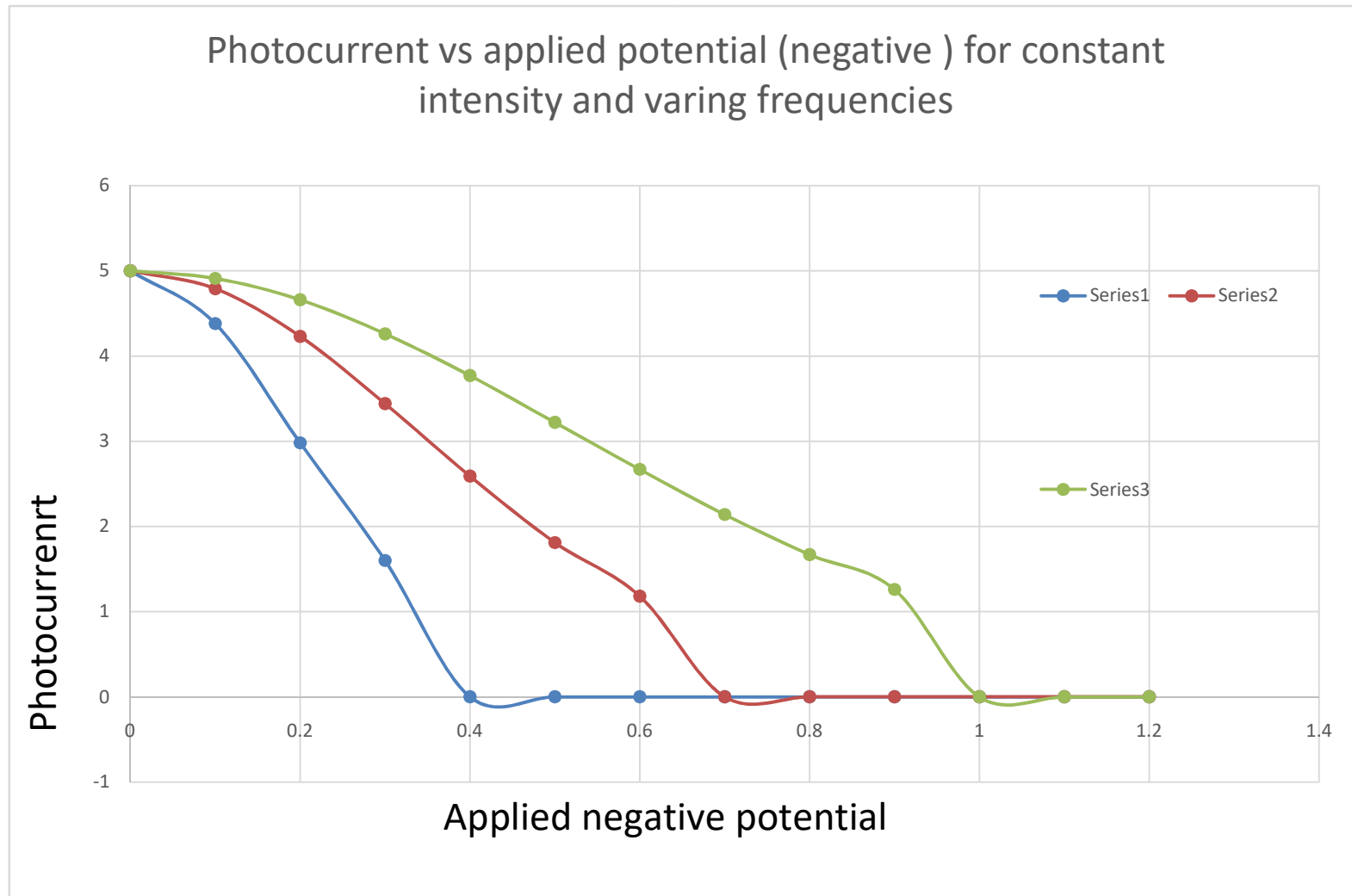
Material- Sodium

Plate area-0.1 cm<sup>2</sup>

Intensity= 10 w/m<sup>2</sup>

S.No.	Voltage (V)	Current ( $\mu$ A)		
		f=6.64 x 10 <sup>14</sup> Hz	f=7.5 x 10 <sup>14</sup> Hz	f=8.57 x 10 <sup>14</sup> Hz
1	0	5	5	5
2	0.1	4.38	4.79	4.91
3	0.2	2.98	4.23	4.66
4	0.3	1.6	3.44	4.26
5	0.4	0	2.59	3.77
6	0.5	0	1.81	3.22
7	0.6	0	1.18	2.67
8	0.7	0	0	2.14
9	0.8	0	0	1.67
10	0.9	0	0	1.26
11	1	0	0	0
12	1.1	0	0	0
13	1.2	0	0	0

# Observations- Graph



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# Observations

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1] Go to the given link and login-

<http://vlab.amrita.edu/index.php?sub=1&brch=195&sim=840&cnt=1>

2] The kinetic energy on decreasing the negative potential at at electron capturing plate— **explained**

3] On increasing intensity at constant frequency, photocurrent increases. —**explained as intensity is related to number of photons per unit area.**

4] On frequency variation at constant intensity, stopping potential increases (more negative potential requires to stop photocurrent) **explained as frequency is related to KE of emitted electrons.**

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# Results

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- (1) The phenomenon of Photoelectric effect is studied as a whole.
- (2) A graph is plotted between photocurrent and applied potential for various intensities at constant frequency.
- (3) A graph is plotted between photocurrent and applied potential for various frequencies at constant intensity.

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# Precautions

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## **Precautions (for physical experiment)-**

1. The wires should be connected tightly.
2. Switch off the various appliances when not in use.

## **Precautions (for virtual experiment)-**

1. Sign up in the given link
1. Cite the reference of virtual link in the project/record of lab work.



THANK YOU

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